Human Phys PCB4701

Lecture 22: Respiration part 2

T. Houpt, Ph.D. houpt@bio.fsu.edu

Hemoglobin and Hemoglobin Dissociation Curves

Transport of CO₂

Respiratory Challenge: Fetus In Utero (in the uterus)

Hemoglobin

Respiratory pigment that picks up oxygen from lungs, releases oxygen into tissues: increases oxygen carrying capacity of blood

Consists of 4 protein subunits (2 x alpha, 2 x beta subunits) and 4 heme molecules (centered around iron atom that binds O₂ molecule). Iron gives blood reddish hue.

oxyhemoglobin: when carrying oxygen; deoxyhemoglobin: when not carrying oxygen.

Dissociation curve: proportion of hemoglobin that is carrying oxygen as a function of oxygen concentration in the blood (or as a function of pH, or temperature, etc.)

Hemoglobin associates with oxygen at high oxygen concentrations (in lungs); oxygen dissociates from hemoglobin at low oxygen concentrations (in tissues).

Bohr Shift: oxygen dissociates from hemoglobin at low pH (reflects high CO₂ in tissues).

2,3 DPG: molecule that is produced when oxyhemoglobin low; causes more oxygen release from hemoglobin.

hemo- blood *globin* - protein

Dissolved O₂ Content of Blood

Henry's Law: $[O_2]$ ml/L = 0.031 x P_{O_2}

at P_{O2} in lungs = 100 mmHg

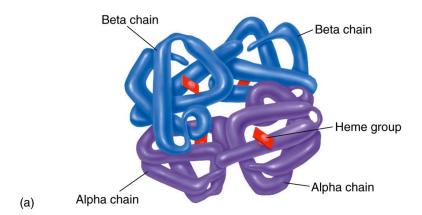
 $[O_2]$ in blood = 3.0ml/L

So to find total dissolved oxygen in human blood (5 L)

= 3.0 ml/L \times 5 L blood = 150 ml O₂ (dissolved)

but human uses 250 ml O₂ per minute....

Hemoglobin



(b)

Figure 16.32

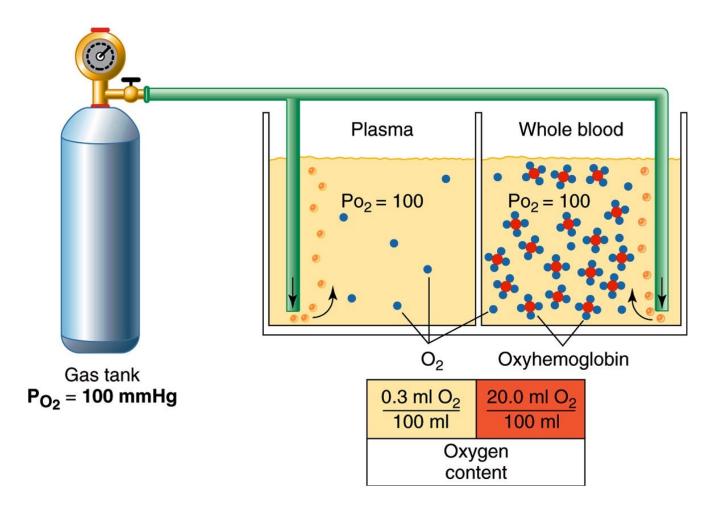
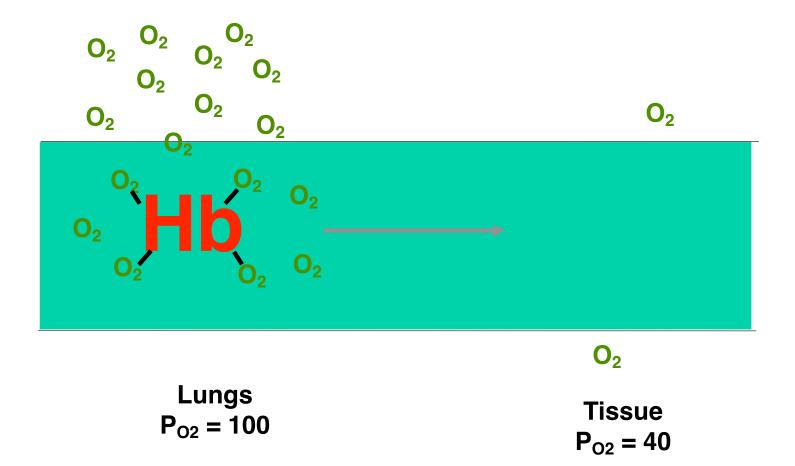
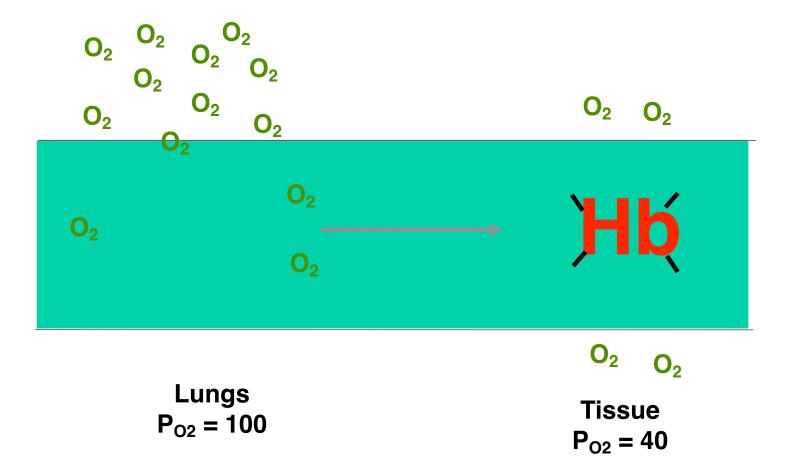


Figure 16.31





O₂ Content of Oxyhemoglobin in Blood

 $[O_2]$ bound to saturated oxyhemoglobin = 1.36 ml/g Hb

[Hb] = 150 g / L of blood

So to find total O_2 in human blood (5 L)

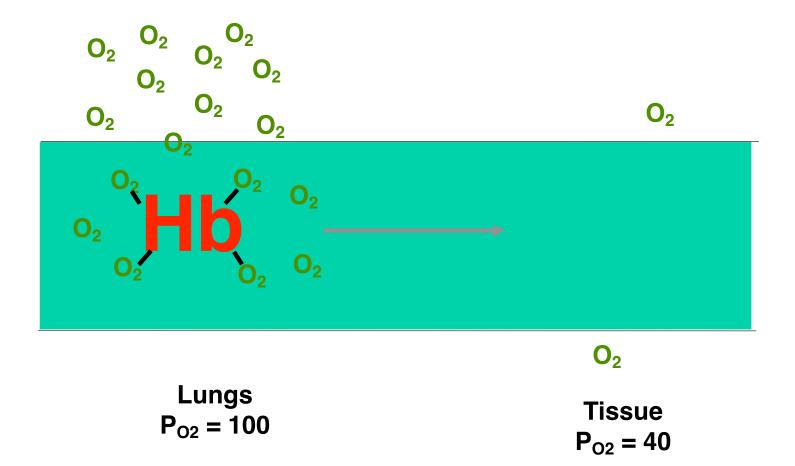
 $I.36 \text{ ml } O_2/\text{ g Hb} \times I50 \text{ g Hb} / L \times 5 \text{ L blood}$

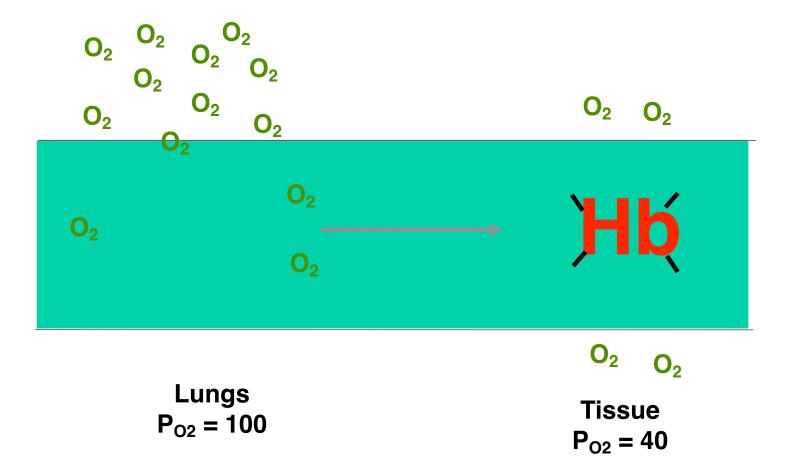
 $= 1020 \text{ ml } O_2$

Total $O_2 = 150$ ml dissolved $O_2 + 1020$ ml O_2 -Hb = 1170 ml

human uses 250 ml O₂ per minute;

so hemoglobin provides a large reserve of O_2





Hemoglobin Dissociation curve

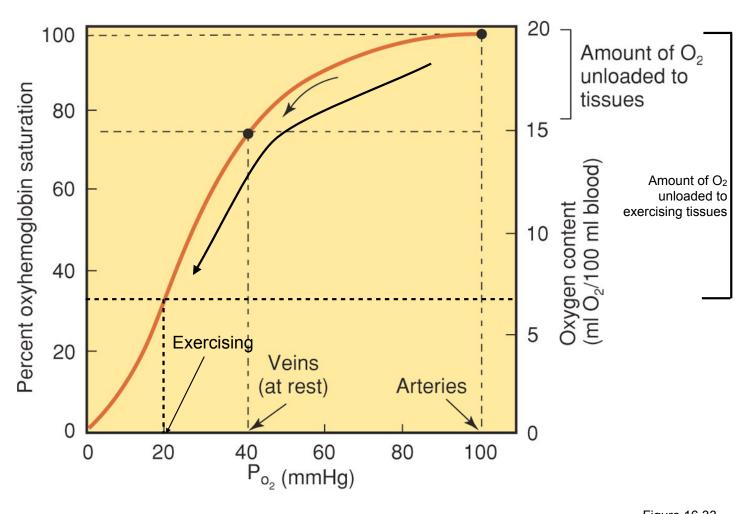


Figure 16.33

Bohr Shift: oxygen dissociates from hemoglobin at low pH (high CO₂ in tissues).

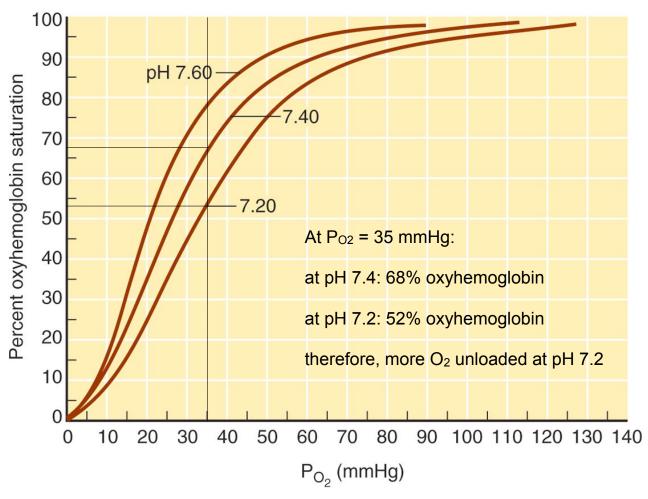


Figure 16.34

Myoglobin

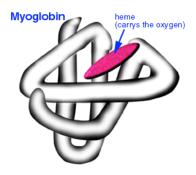
Respiratory pigment in skeletal and cardiac muscle cells. Similar to hemoglobin, but only 1 heme molecule. Stores oxygen in muscle cells, helps transfer oxygen from blood to muscle mitochondria.

Dissociation curve for myoglobin is shifted to the left: only unloads oxygen at very low oxygen levels (during exercise).

In heart muscle, myoglobin stores oxygen for use during systole, when coronary arteries are squeezed shut by contraction.

Myoglobin in the blood indicates damage to muscle tissue.

Myoglobin



Hemoglobin

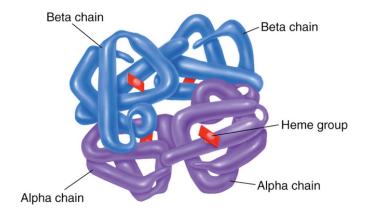
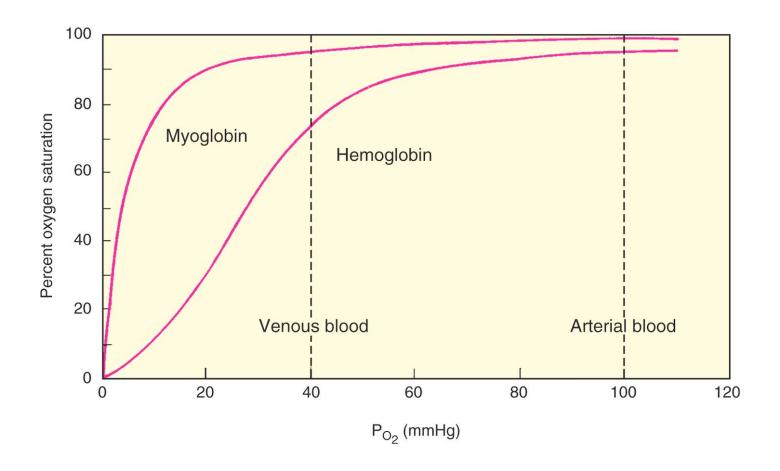


Figure 16.37 Myoglobin optimized for Low Oxygen Conditions



Carbon Dioxide transport

Carbon Dioxide is transported in the blood as:

dissolved CO₂ (7% of blood CO₂) carbaminohemoglobin (20% of blood CO₂) bicarbonate ion (HCO₃⁻)

In Tissue Capillaries: Carbonic Anhydrase in red blood cells catalyzes H₂CO₃ formation. H+ binds to deoxyhemoglobin, HCO₃⁻ is exchanged for Cl- ion from plasma (chloride shift)

In Pulmonary Capillaries: Reverse chloride shift exchanges CI- ion to bring HCO3-into red blood cell. Oxyhemoglobin releases H+, so H₂CO₃ -> CO₂ is formed and diffuses into lungs.

Carbon Monoxide poisoning

CO also binds hemoglobin, but binding is **irreversible** (210x higher than O₂). Binding of 50-80% of hemoglobin is fatal. Victims of CO poisoning are cherry red, because of carboxyhemoglobin color.

Carbon Dioxide transport

Carbon Dioxide is transported in the blood as:

dissolved CO₂ (7% of blood CO₂) carbaminohemoglobin (20% of blood CO₂) bicarbonate ion (HCO₃⁻)

In Tissue Capillaries: Carbonic Anhydrase in red blood cells catalyzes H₂CO₃ formation. H+ binds to deoxyhemoglobin, HCO₃⁻ is exchanged for Cl- ion from plasma (chloride shift)

In Pulmonary Capillaries: Reverse chloride shift exchanges CI- ion to bring HCO3-into red blood cell. Oxyhemoglobin releases H+, so H₂CO₃ -> CO₂ is formed and diffuses into lungs.

Carbon Monoxide poisoning

CO also binds hemoglobin, but binding is **irreversible** (210x higher than O₂). Binding of 50-80% of hemoglobin is fatal. Victims of CO poisoning are cherry red, because of carboxyhemoglobin color.

Figure 16.38

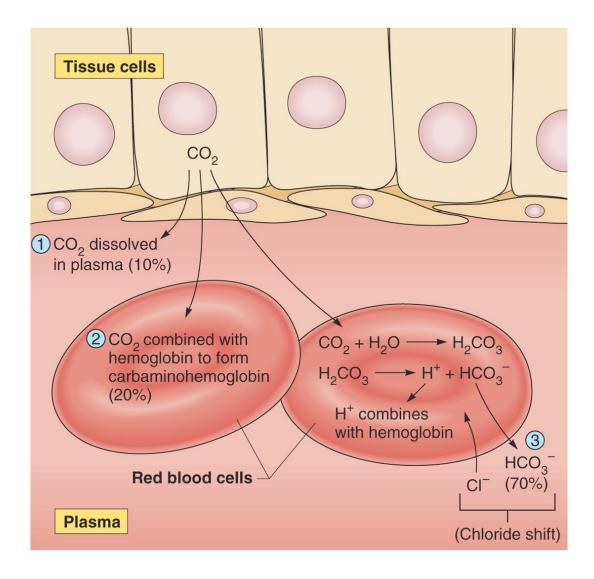
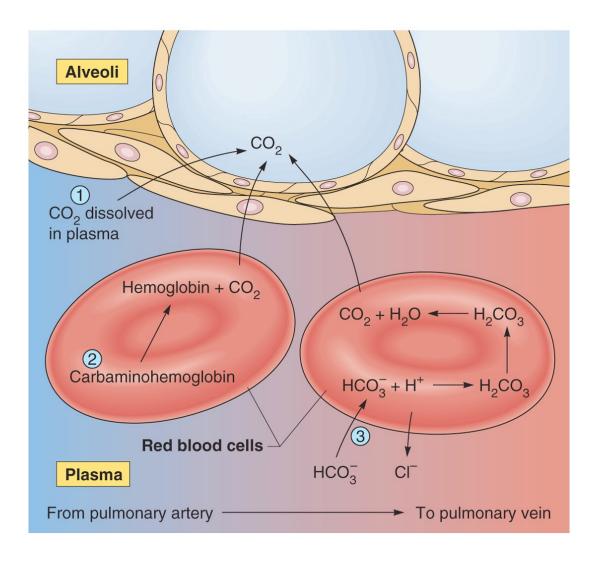
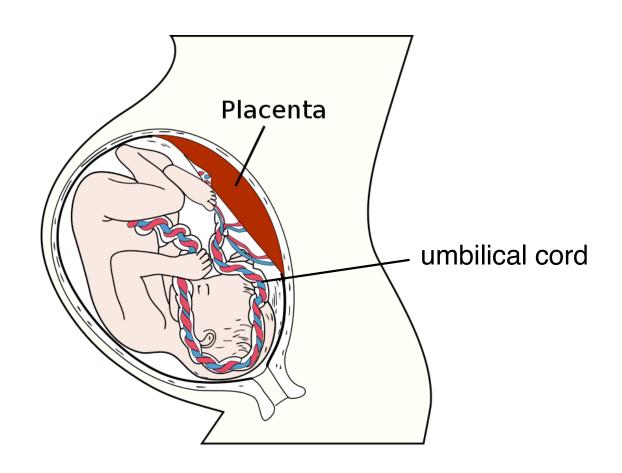


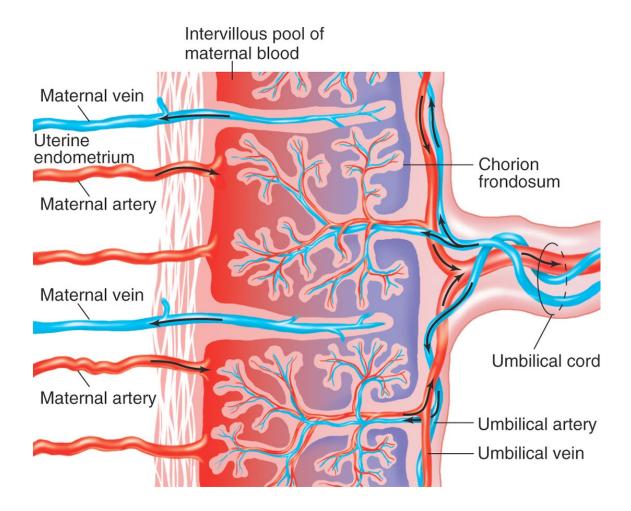
Figure 16.39



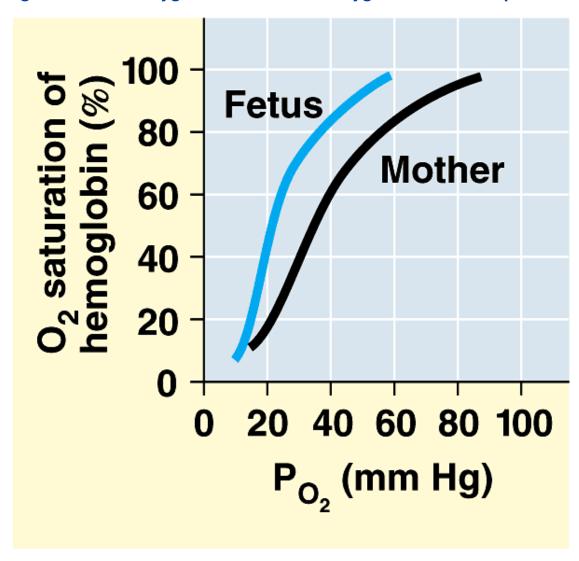
Fetal Circulation

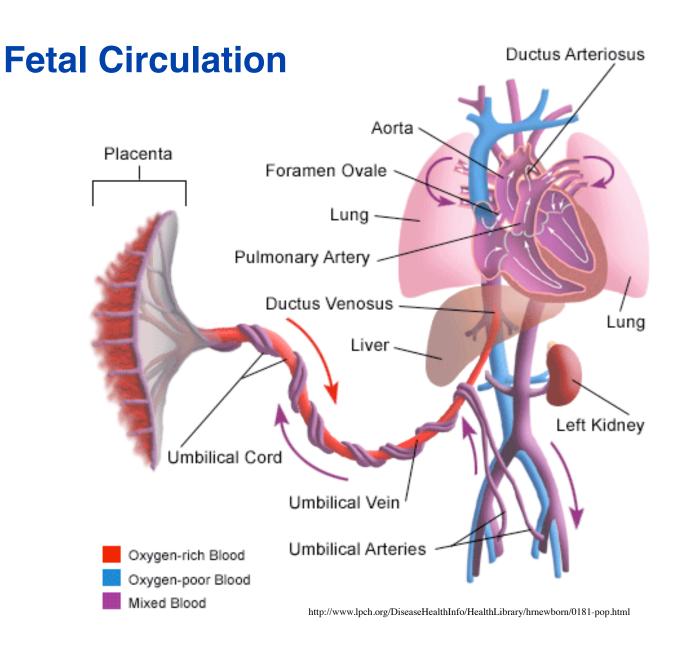
- Placenta: Gas (and nutrient) exchange in high-surface area capillary bed between mom and fetus
- Fetal Hemoglobin binds more O₂ in low-O₂ environment of placenta
- Circulation of blood through umbilical cord to placenta to exchange gases.
- Shunts in fetal heart to send O₂ rich blood directly from vena cava to left atrium (oval foramen) and to aorta (ductus arteriosus) so that lungs are bypassed

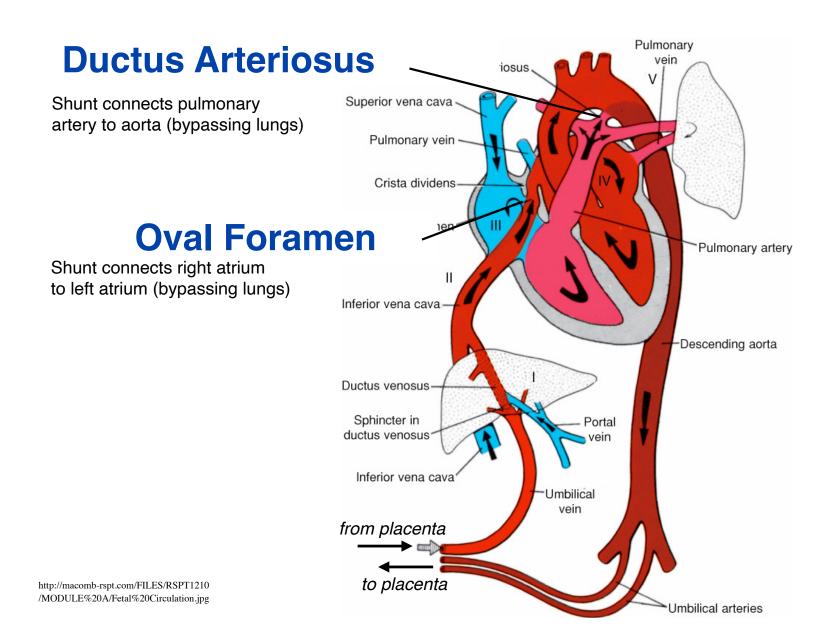




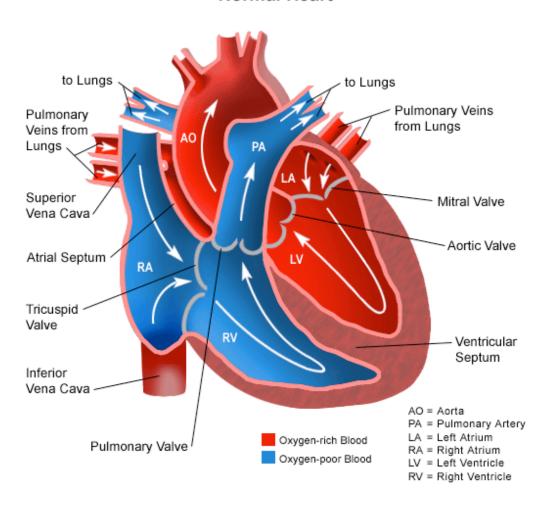
Fetal Hemoglobin: binds oxygen better in lower oxygen condition of placenta





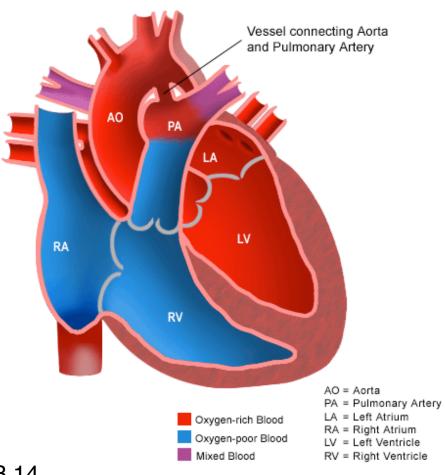


Normal Heart



http://www.chop.edu/healthinfo/patent-ductus-arteriosus-pda.html

Patent Ductus Arteriosus (PDA)



see Figure 13.14

http://www.chop.edu/healthinfo/patent-ductus-arteriosus-pda.html